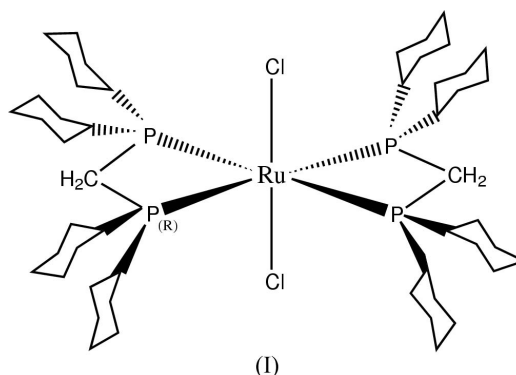


**(*Trans*-dichloro- $\kappa$ Cl)bis[(dicyclohexylphosphino)methane- $\kappa^2$ P,P']ruthenium(II)**Linfang Jia,<sup>a</sup> Wenfu Fu,<sup>a,b\*</sup>  
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**Key indicators**Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å  
 $R$  factor = 0.032  
 $wR$  factor = 0.083  
Data-to-parameter ratio = 18.2For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.In the title complex,  $[\text{Ru}(\text{C}_{25}\text{H}_{46}\text{P}_2)_2\text{Cl}_2]$  or  $\text{RuCl}_2(\text{dcpm})_2$  [dcpm is 1,2-bis(dicyclohexylphosphino)methane,  $\text{C}_{25}\text{H}_{46}\text{P}_2$ ], the Ru atom (site symmetry  $\bar{1}$ ) is in an octahedral coordination environment with two chelating dcpm ligands and two *trans* chloro ligands.**Comment**Ruthenium complexes containing chelating phosphine ligands play a prominent role as active catalysts in many fields (Collman, 1987; White & Coville, 1994). Many of these complexes have been prepared and characterized (Shawkatally *et al.*, 1998; Lucas *et al.*, 2000; Hockless *et al.*, 1996), but complexes containing dcpm [dcpm is 1,2-bis(dicyclohexylphosphino)methane,  $\text{C}_{25}\text{H}_{46}\text{P}_2$ ] are uncommon (Joslin *et al.*, 1991). Here, we report the synthesis and structure of the title compound, (I), (Fig. 1), which may serve as a useful precursor for the synthesis of further ruthenium complexes (Grocott & Wild, 1982; Nolan *et al.*, 1997; Volland *et al.*, 2004).In (I) the central Ru(II) atom occupies an inversion centre and is coordinated by two *trans* chloro ligands and two *P,P'*-chelating dcpm ligands, resulting in octahedral geometry for the ruthenium atom (Table 1). The Ru–Cl distances in (I) are comparable to those in related compounds (Blake *et al.*, 1993; Hockless *et al.*, 1996). The two Ru–P distances in (I) differ by 0.056 Å. The packing in (I) is shown in Fig. 2.**Experimental**Compound (I) was prepared from the reaction of the chelating phosphine ligand with ruthenium(III) chloride by a standard procedure (Grocott & Wild, 1982). Suitable crystals for single-crystal analysis were grown by diffusion of diethyl ether into a dichloromethane solution of (I) at room temperature. Yield 85%. Elemental analysis calcd for  $\text{C}_{50}\text{H}_{92}\text{Cl}_2\text{P}_4\text{Ru}$ : C 60.72%, H 9.37%, found: C 60.68%, H 9.35%.

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## Crystal data

[Ru(C<sub>25</sub>H<sub>46</sub>P<sub>2</sub>)<sub>2</sub>Cl<sub>2</sub>]  
*M<sub>r</sub>* = 989.09  
 Monoclinic, *C*2/*c*  
*a* = 24.120 (11) Å  
*b* = 10.732 (5) Å  
*c* = 21.780 (11) Å  
 $\beta$  = 108.098 (8)°  
*V* = 5359 (4) Å<sup>3</sup>  
*Z* = 4

*D<sub>x</sub>* = 1.226 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 Cell parameters from 858 reflections  
 $\theta$  = 2.7–26.3°  
 $\mu$  = 0.54 mm<sup>-1</sup>  
*T* = 293 (2) K  
 Block, orange  
 0.30 × 0.18 × 0.12 mm

## Data collection

Bruker SMART CCD diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 1998)  
*T<sub>min</sub>* = 0.796, *T<sub>max</sub>* = 0.937  
 13674 measured reflections

4717 independent reflections  
 3644 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.040  
 $\theta_{\max}$  = 25.0°  
*h* = -28 → 24  
*k* = -12 → 12  
*l* = -25 → 21

## Refinement

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.032  
*wR*(*F*<sup>2</sup>) = 0.083  
*S* = 1.09  
 4717 reflections  
 259 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0366P)^2 + 1.6382P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.30 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.42 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Ru1–P2	2.3669 (12)	Ru1–Cl1	2.4334 (11)
Ru1–P1	2.3775 (10)		
P2–Ru1–P1	71.42 (4)	P2–Cl13–P1	97.10 (13)

H atoms were initially located in a difference Fourier map, then placed in ideal positions, with C–H = 0.97 Å, and refined as riding, with the constraint *U<sub>iso</sub>*(H) = 1.2*U<sub>eq</sub>*(C) applied.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART* or *SAINT*?; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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## References

- Blake, A. J., Fotheringham, J. D. & Stephenson, T. A. (1993). *Acta Cryst.* **C49**, 142–144.  
 Bruker (1998). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Collman, J. P., Hegedus, L. S., Norton, J. R. & Finke, R. G. (1987). *Principles and Applications of Organotransition Metal Chemistry*, 2nd ed. Mill Valley, CA: University Science.  
 Grocott, S. C. & Wild, S. B. (1982). *Inorg. Chem.* **21**, 3527–3534.  
 Hockless, D. C. R., Wild, S. B., McDonagh, A. M., Whittall, I. R. & Humphrey, M. G. (1996). *Acta Cryst.* **C52**, 1639–1641.

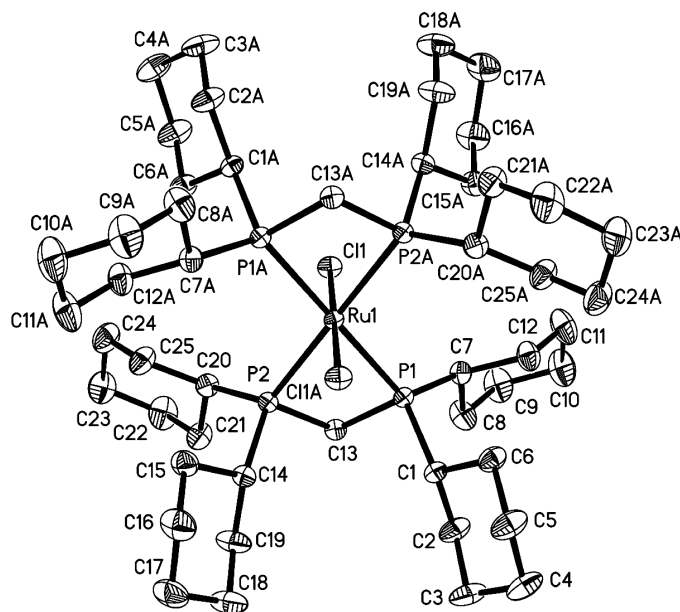


Figure 1

View of (I), with 30% displacement ellipsoids and H atoms omitted for clarity. Atoms with the suffix A are generated by the symmetry operation  $1 - x, 2 - y, -z$ .

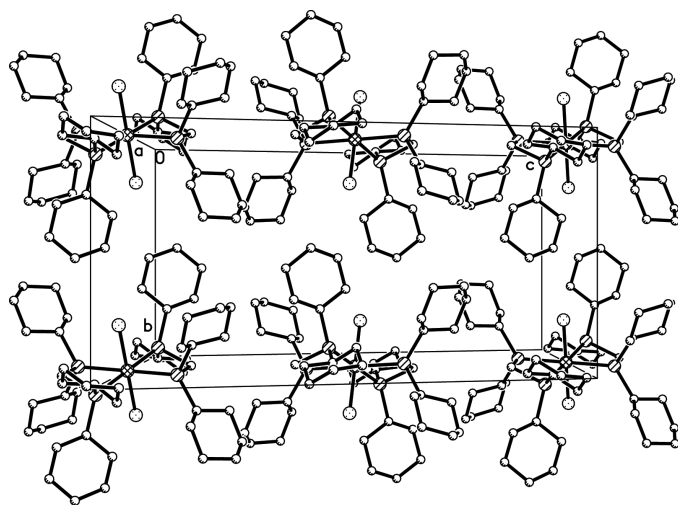


Figure 2

Unit-cell packing in (I), viewed along the *a* axis. H atoms have been omitted for clarity.

- Joslin, F. L., Mague, J. T., Roundhill, D. M. (1991). *Polyhedron*, **10**, 1713–1715.  
 Lucas, N. T., Powell, C. E. & Humphrey, M. G. (2000). *Acta Cryst.* **C56**, e392–e393.  
 Nolan, S. P., Belderrain, T. R. & Grubbs, R. H. (1997). *Organometallics*, **16**, 5569–5571.  
 Shawkataly O. B., Ramalingam K., Ashari D. M., Fun H.-K. & Razak I. A. (1998). *Acta Cryst.* **C54**, 329–331.  
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). *SHELXTL*. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Volland M. A. O., Hansen S. M., Rominger F. & Hofmann P. (2004). *Organometallics*, **23**, 800–816.  
 White D., Coville N. (1994). *Adv. Organomet. Chem.* **36**, 95–158.